

## SESQUITERPENE LACTONES FROM *BRACHYLAENA* SPECIES\*

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(Received 8 April 1981)

**Key Word Index**—*Brachylaena transvaalensis*; *B. rotundata*; Compositae; sesquiterpene lactones; germacranolides; guaianolides; eudesmanolides; chemotaxonomy.

**Abstract**—The investigation of two *Brachylaena* species afforded in addition to known sesquiterpene lactones and other constituents five new lactones, a germacranolide, a guaianolide and three eudesmanolides. The structures were elucidated by spectroscopic methods. The chemotaxonomy of this complex genus is discussed.

### INTRODUCTION

The Tarchonantheae have been excluded from the tribe Inuleae (Compositae) following detailed taxonomic studies [1–4]. However, no clear decision was made where to place this group. So far, little is known on the chemistry of the three genera *Tarchonanthus*, *Brachylaena* and *Synchodendron* belonging to this subtribe. The essential oil of *Brachylaena hutchinsii* afforded several sesquiterpenes, mainly cadinene and copaene derivatives [5, 6], while from *Tarchonanthus* species an acetylenic thiophene and also some simple sesquiterpenes were reported [7]. We have now investigated two *Brachylaena* species, which afforded several sesquiterpene lactones, whose occurrence may be useful for the placement of this genus.

### RESULTS AND DISCUSSION

The aerial parts of *Brachylaena transvaalensis* Hutch. ex Phill. et Schweick, afforded the acetylenic compounds **1** and **2**, germacrene D, lupeyl acetate, linolic and linolenic acid, the germacranolides **6a** [8], **6b** [8], onopordopicrin (**7**) [9], salonitenolide (**9**) [10] and two further lactones, the epoxy derivative of **6b** (**8a**) as well as a dihydro derivative of zaluzanin C, the guaianolide **14a**. The structure of **8a**, which on acetylation afforded the acetate **8b**, followed from the molecular formula and the <sup>1</sup>H NMR data (Table 1). Though some signals were overlapping multiplets, comparison with the spectra of **6b** and **7** clearly showed that a germacranolide with oxygen functions at C-8 and C-15 was present. The chemical shifts of H-15 observed after acetylation established the position of the free hydroxyl group, while the characteristic doublets at δ 3.10 and 2.81 as well as the singlet at 1.58 indicated the presence of an epoxy isobutyrate which must be α-orientated at C-8 as the

Table 1. <sup>1</sup>H NMR spectral data of compounds **8a** and **8b** (400 MHz, CDCl<sub>3</sub>, TMS as internal standard)

	<b>8a</b>	<b>8b</b>
H-1	4.98br dd	4.97m
H-2	1.97m	2.0m
H-2'	2.22m	2.22m
H-3	2.56ddd	2.52m
H-3'	2.22m	2.22m
H-5	4.78br d	4.85m
H-6	5.04br dd	
H-7	3.02dddd	3.02m
H-8	5.12br dd	5.09br dd
H-9	2.53br d	2.52m
H-9'	2.41dd	2.42dd
H-13	6.39d	6.40d
H-13'	6.10d	6.10d
H-14	1.48br s	1.58br s
H-15	4.28br d	4.62d
H-15'	4.09br d	4.57d
OCOR	3.10d	3.09d
	2.81d	2.81d
	1.58s	1.58s
OAc	—	2.10s

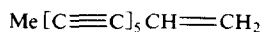
$J(\text{Hz})$ : 1,2 ~ 4; 1,2' = 11; 2,3 = 2,3' ~ 3; 3,3' = 12; 5,6 = 6,7 = 10; 7,8 = 9; 7,13 = 3.5; 7,13' = 3; 8,9' = 10; 9,9' = 13; 3', 3<sub>2</sub> = 6.

coupling  $J_{7,8}$  was 9 Hz as in the spectra of **6a**, **6b**, **7** and **9**. The presence of a 6,12-*trans* lactone followed from the couplings  $J_{5,6}$  and  $J_{6,7}$ . The <sup>1</sup>H NMR data (Table 2) of **14a** and of the acetate **14b**, obtained by mild acetylation, as well as the molecular formula showed that a dihydro derivative of a trimethylene sesquiterpene lactone with a free hydroxyl group must be present. The <sup>1</sup>H NMR data were similar to those of zaluzanin C. However, the H-13 methylene

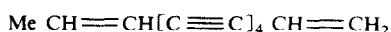
\*Part 378 in the series 'Naturally Occurring Terpene Derivatives'. For Part 377, see Bohlmann, F., Singh, P., Jakupovic, J., Robinson, H. and King, R. M. (1982) *Phytochemistry* **20**, 707.

signals were replaced by a double quartet at  $\delta$  2.23 and a doublet at 1.22, indicating the presence of an 11,13-dihydrozaluzanin C. The stereochemistry at C-11 was deduced from the large coupling  $J_{7,11}$  and the chemical shift of H-13. Spin decoupling supported the assignments of the signals. Irradiation of the downfield signal at 5.54 in the spectrum of the acetate collapsed the H-15 signals to doublets and also changed the H-2 signals in the expected way. Irradiation of the double doublet of the proton under the lactone oxygen (H-6) allowed the assignment of H-5 and H-7. All other signals were assigned in the usual way. The roots afforded 1-3, lupeyl acetate, its  $\Delta^{12}$ -isomer and 9-oxo-nerolidol (4), which was identical

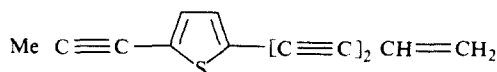
with a ketone obtained by oxidation of 9-hydroxy-nerolidol isolated from an *Artemisia* species [11]. Furthermore, a complex mixture of several sesquiterpene lactones was present, which could be separated only with difficulty. Finally 12 lactones were isolated, which included costunolide (5), dehydrocostuslactone (10) [12], dehydrozaluzanin C (11) [13], zaluzanin C (12) [13], dihydrodehydrocostuslactone (13) [14], 4 $\beta$ , 15-dihydrodehydrozaluzanin C (15) [15], tetrahydrodehydrozaluzanin C (16) [15], the furanoheliangolide 17 [16], tubiferin (18) [17] and three further eudesmanolides.  $^1\text{H}$  NMR investigations of these lactones, which were available only in minute amounts, led to structures 19-21 (Table 3).



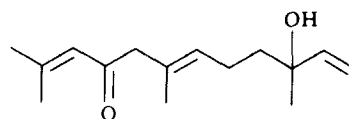
1



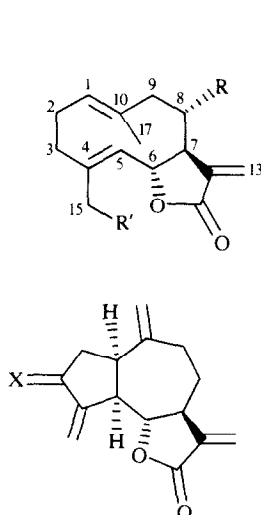
2



3



4



5 R = R' = H

6aR = OiBu, R' = OH

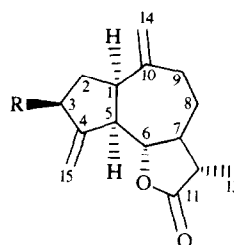
6bR = OMeacr, R' = OH

7 R = O-C(=O)-CH(OH)-CH<sub>3</sub>, R' = OH8aR = O-C(=O)-C(CH<sub>3</sub>)<sub>2</sub>-O, R' = OH8bR = O-C(=O)-C(CH<sub>3</sub>)<sub>2</sub>-O, R' = OAc

9 R = R' = OH

10 X = H<sub>2</sub>

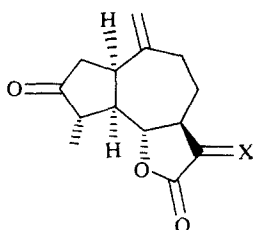
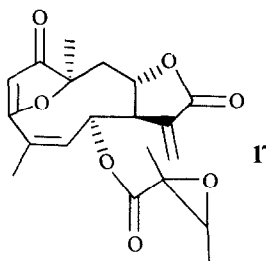
11 X = O

12 X = H,  $\beta$ OH

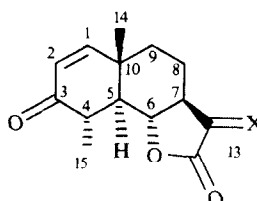
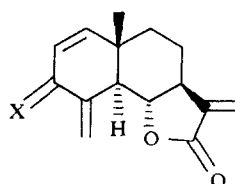
13 R = H

14a R = OH

14b R = OAc

15 X = CH<sub>2</sub>16 X = H,  $\alpha$  Me

17

18 X = CH<sub>2</sub>19 X = H,  $\alpha$  Me20 X = H,  $\alpha$  OH

21 X = O

Table 2. <sup>1</sup>H NMR spectral data of compounds **14a** and **14b** (400 MHz, CDCl<sub>3</sub>, TMS as internal standard)

	<b>14a</b>	<b>14b</b>
H-1	2.94br ddd	2.90br ddd
H-2	2.54ddd	2.46ddd
H-2'	1.75ddd	1.79ddd
H-3	4.54br dd	5.54ddd
H-5	2.78ddd	2.80ddd
H-6	4.03dd	3.99dd
H-7	1.89ddd	1.91ddd
H-8	2.13ddd	2.12ddd
H-8'	1.34m	1.31br dd
H-9	2.34ddd	2.51ddd
H-9'	2.02ddd	2.03ddd
H-11	2.23dq	2.23dq
H-13	1.22d	1.22d
H-14	4.96br s	4.92br s
H-14'	4.93br s	4.90br s
H-15	5.39dd	5.41dd
H-15'	5.30dd	5.27dd
OAc	—	2.10s

*J*(Hz): 1,2 = 1,2' = 1,5 ~ 9; 2,2' = 14; 2,3 = 2,3' = 7.5; 3,15 = 5,15 = 2; 5,6 = 6,7 = 10; 7,8 = 4; 7,8' = 11; 7,11 = 11.5; 8,8' = 13; 8,9 = 8,9' ~ 4; 8,9' = 4; 8',9' = 10; 11,13 = 7.

Table 3. <sup>1</sup>H NMR spectral data of compounds **19–21** (400 MHz, CDCl<sub>3</sub>, TMS as internal standard)

	<b>19</b>	<b>20</b>	<b>21</b>
H-1	6.69d	5.62dd	6.81d
H-2	5.88d	5.56dd	6.03d
H-3	—	4.72br	—
H-4	2.57dq	—	—
H-5	1.96dd	2.60br d	3.03ddd
H-6	3.98dd	4.46dd	4.13dd
H-7	1.63m	2.58dddd	2.63dddd
H-8 $\alpha$	1.90m	2.18ddd	2.14m
H-8 $\beta$	1.74m	1.64ddd	1.7m
H-9 $\alpha$	—	1.56m	1.38m
H-9 $\beta$	1.60m	1.73ddd	1.73m
H-11	2.28dq	—	—
H-13	1.21d	6.10d	6.14d
H-13'	—	5.40br s*	5.46d
H-14	1.15s	0.92s	1.05s
H-15	1.34d	5.40br s*	6.28dd
H-15'	—	5.04br t	5.71br s

\*CHCl<sub>3</sub>-C<sub>6</sub>D<sub>6</sub> H-13' 5.24d, H-15 5.34br d.

*J*(Hz): compound **19**: 1,2 = 10; 4,5 = 12; 4,15 = 7; 5,6 = 11; 6,7 = 10; 7,11 = 12; 11,13 = 7; compound **20**: 1,2 = 10; 1,3 = 2,3 ~ 2; 3,15 = 1.5; 5,6 = 11; 6,7 = 10.5; 7,8 $\alpha$  = 3; 7,8 $\beta$  = 12; 7,13 = 3.3; 7,13' = 3; 7,8 $\alpha$  = 3; 7,8 $\beta$  = 12; 8 $\alpha$ ,8 $\beta$  = 13; 8 $\alpha$ ,9 $\alpha$  = 8 $\alpha$ ,9 $\beta$  ~ 3; 8 $\beta$ ,9 $\alpha$  = 11; 8 $\beta$ ,9 $\beta$  ~ 3; 9 $\alpha$ ,9 $\beta$  = 13; compound **21**: 1,2 = 10; 5,6 = 6,7 = 11; 5,15 ~ 2; 7,8 $\alpha$  ~ 3; 7,8 $\beta$  ~ 10; 7,13 = 3.5; 7,13' = 3.

Compound **19** could be separated from **18** only by transforming the latter to the corresponding pyrazoline derivative. The <sup>1</sup>H NMR spectral data clearly showed the presence of one tertiary and two secondary methyls. Furthermore, the location of the conjugated keto group followed from the low-field doublets. A double doublet at 3.98 was obviously that of the hydrogen under the lactone oxygen (H-6). Irradiation of this signal allowed assignment of the signals of H-5 and H-7. Their irradiation established the signals of H-4, H-8 and H-11. As the H-4 signal, which was coupled with the methyl group, was at relatively low field, a neighbouring keto group was required, thus establishing the whole sequence from C-1 to C-9 and therefore revealing the presence of an eudesmanolide. The stereochemistry at C-4 to C-7 was deduced from the couplings observed. The large coupling *J*<sub>4,5</sub> and *J*<sub>7,11</sub> supported an  $\alpha$ -orientation of the methyls at C-4 and C-11. The structure of **20** could also be deduced from the <sup>1</sup>H NMR data (Table 3). The position of the hydroxyl group followed from the splitting of the olefinic signals and from the result of decoupling experiments. Irradiation of the low-field broadened signal at 4.72 collapsed the olefinic signals to doublets and sharpened the signals of the exomethylene protons. As these signals were also coupled with a broadened doublet at 2.60, which was further coupled with the lactone proton, the assignments of H-1, H-2, H-3, H-5, H-6 and H-15 were possible. The remaining signals could be assigned by further decouplings. Inspection of a model supported the proposed  $\alpha$ -orientation of the hydroxyl group at C-3, as an  $\alpha$ -proton at C-3 should show a larger coupling *J*<sub>2,3</sub>. We have named compound **20** brachylaenolide.

Having established the structures of **19** and **20**, the structure of **21** could be deduced from the <sup>1</sup>H NMR data by comparison with those of **18–20**. As expected, the signals of H-1, H-2, H-5 and H-15 were shifted downfield and one of the H-15 signals was a clear double doublet, caused by an allylic and a geminal coupling as could be established by spin decoupling. The structures were further established by oxidation of **20**, which gave a ketone identical with the natural compound.

The aerial parts of *Brachylaena rotundata* S. Moore also afforded the germacranolides **6a**, **6b** and **8a** as well as lupeyl acetate and its  $\Delta^{12}$ -isomer.

The constituents isolated from the two species may support the results on pollen morphology [2–4], which showed relationships of the Tarchonantheae to either Mutisieae or Anthemideae. Placement in the latter tribe seems very unlikely as the acetylenes present in both *Brachylaena* and *Tarchonanthus* have never been detected in any member of the Anthemideae. However, a relationship to the Mutisieae is supported by the constituents found in the *Brachylaena* spp. In particular, the lactones isolated from genera belonging to the subtribe Gochnatinae are in part similar to those of *Brachylaena* (*Actinoseris* [18], *Cnicothamnus* [19], *Dicoma* [20], *Gochnatia* [19, 21], *Moquinia* [22], *Pertya* [23] and *Wunderlichia* [18, 24]). A few lactones were also isolated from other subtribes of the Mutisieae. Acetylenes of the type 1–3 have also been isolated from the Mutisieae; but they are widespread in other plants. Also, sesquiterpene lac-

tones of the types described here have been isolated from members of other tribes. Clearly, more species have to be investigated to obtain a precise picture of the relationships of this group.

### EXPERIMENTAL

The air-dried plant material, collected in Transvaal, was extracted with Et<sub>2</sub>O–petrol (1:2) and the resulting extracts were separated first by column chromatography (Si gel) and further by repeated TLC (Si gel). Known compounds were identified by comparing their IR and <sup>1</sup>H NMR spectra with those of authentic material. The voucher specimens are deposited in the National Botanic Research Institute, Pretoria.

*Brachylaena transvaalensis* Hutch. ex Phill. et Schweick (voucher 81/36). The roots (260 g) afforded 100 mg lupeyl acetate and its Δ<sup>12</sup>-isomer (ca 2:1), 5 mg germacrene D, 0.2 mg **1**, 1 mg **2**, 1 mg **3**, 5 mg **4**, 5 mg **5**, 2 mg **10**, 2 mg **11**, 4 mg **12**, 5 mg **13**, 2 mg **15**, 5 mg **16**, 10 mg **17**, 4 mg **18**, 3 mg **19**, 2 mg **20** and 2 mg **21**. The aerial parts (300 g) gave 40 mg linolic and linolenic acid (1:1), 2 mg germacrene D, 5 mg lupeyl acetate, 0.1 mg **1**, 0.2 mg **2**, 2 mg **6a**, 2 mg **6b**, 10 mg **7**, 3 mg **8a**, 5 mg **9** and 2 mg **14a**. The lactones were separated by repeated TLC using first Et<sub>2</sub>O–petrol mixtures and then mixtures of CHCl<sub>3</sub>–C<sub>6</sub>H<sub>6</sub>–Et<sub>2</sub>O.

*Brachylaena rotundata* (voucher 81/169). The aerial parts (50 g) afforded 50 mg lupeyl acetate and its Δ<sup>12</sup>-isomer (2:1), 1 mg **6a**, 3 mg **6b** and 1 mg **8a**.

*Salonitenolide-8-O-2,3-epoxy isobutyrate* (**8a**). Colourless gum, IR ν<sub>max</sub><sup>CCl<sub>4</sub></sup> cm<sup>-1</sup>: 3600 (OH), 1780 (lactone), 1740 (CO<sub>2</sub>R); MS *m/z* (rel. int.): 246.125 [M – RCO<sub>2</sub>H]<sup>+</sup> (3) (C<sub>15</sub>H<sub>18</sub>O<sub>3</sub>), 228 [246 – H<sub>2</sub>O]<sup>+</sup> (9), 215 [246 – CH<sub>2</sub>O]<sup>+</sup> (20), 119 (68), 91 (100). Compound **8a** (3 mg) was heated in 0.5 ml Ac<sub>2</sub>O for 30 min at 70°. TLC (Et<sub>2</sub>O–petrol, 3:1) afforded 2 mg **8b**, colourless gum (for <sup>1</sup>H NMR spectrum see Table 1).

11β,13-Dihydrozaluzanin C (**14a**). Colourless gum, which was purified as its acetate **14b** (heating in Ac<sub>2</sub>O for 1 hr), colourless gum, IR ν<sub>max</sub><sup>CCl<sub>4</sub></sup> cm<sup>-1</sup>: 1790 (lactone), 1745, 1250 (OAc); MS *m/z* (rel. int.): 248.141 [M – ketene]<sup>+</sup> (62) (C<sub>15</sub>H<sub>20</sub>O<sub>3</sub>), 247 [M – MeCO]<sup>+</sup> (63), 230 [M – AcOH]<sup>+</sup> (27), 157 (70), 156 (67), 91 (78), 55 (100).

$$[\alpha]_{24^\circ}^{CHCl_3} = \frac{589 \quad 578 \quad 546 \quad 436 \quad 365 \text{ nm}}{+11 \quad +14 \quad +17 \quad +26 \quad +41} \quad (CHCl_3; c0.23).$$

11β,13-Dihydrotuberiferin (**19**). Separated from **18** by addition of CH<sub>3</sub>N<sub>2</sub>, which transformed **18** to the pyrazoline derivative. TLC (Et<sub>2</sub>O–petrol, 3:1) afforded pure **19**. Colourless crystals, mp 136° (Et<sub>2</sub>O–petrol), IR ν<sub>max</sub><sup>CCl<sub>4</sub></sup> cm<sup>-1</sup>: 1790 (lactone), 1685 (C=CCO); MS *m/z* (rel. int.): 248.141 [M]<sup>+</sup> (29) (C<sub>15</sub>H<sub>20</sub>O<sub>3</sub>), 220 [M – CO]<sup>+</sup> (30), 205 [220 – Me]<sup>+</sup> (11), 192 [220 – CO]<sup>+</sup> (23), 173 (78), 69 (100).

$$[\alpha]_{24^\circ}^{CHCl_3} = \frac{589 \quad 578 \quad 546 \quad 436 \quad 365 \text{ nm}}{+27 \quad +13 \quad 0 \quad -30 \quad -300} \quad (CHCl_3; c0.03).$$

*Brachylaenolide* (**20**). Colourless crystals, mp 138° (Et<sub>2</sub>O–petrol), IR ν<sub>max</sub><sup>CCl<sub>4</sub></sup> cm<sup>-1</sup>: 3600 (OH), 1790 (lactone); MS *m/z* (rel. int.): 246.125 [M]<sup>+</sup> (54) (C<sub>15</sub>H<sub>18</sub>O<sub>3</sub>), 231 [M – Me]<sup>+</sup> (18), 217 [M – CHO]<sup>+</sup> (28), 203 [231 – CO]<sup>+</sup> (13), 173 (80), 91 (94), 55 (100).

$$[\alpha]_{24^\circ}^{CHCl_3} = \frac{589 \quad 578 \quad 546 \quad 436 \quad 365 \text{ nm}}{+150 \quad +158 \quad +175 \quad +305 \quad +493} \quad (CHCl_3; c0.04).$$

Compound **20** (2 mg) in 2 ml Et<sub>2</sub>O was stirred with 30 mg MnO<sub>2</sub>. TLC (Et<sub>2</sub>O–petrol, 3:1) afforded 1 mg **21**, identical with the natural compound.

*Dehydrobrachylaenolide* (**21**). Colourless crystals, mp 225° (iso-PrOH), IR ν<sub>max</sub><sup>CHCl<sub>3</sub></sup> cm<sup>-1</sup>: 1775 (lactone), 1680, 1620 (C=CCO); MS *m/z* (rel. int.): 244.110 [M]<sup>+</sup> (43), 229 [M – Me]<sup>+</sup> (11), 216 [M – CO]<sup>+</sup> (23), 215 [M – CHO]<sup>+</sup> (25), 201 [229 – CO]<sup>+</sup> (15), 91 (100).

$$[\alpha]_{24^\circ}^{CHCl_3} = \frac{589 \quad 578 \quad 546 \quad 436 \quad 365 \text{ nm}}{+67 \quad +67 \quad +18 \quad -115 \quad +365} \quad (CHCl_3; c0.16).$$

**Acknowledgements**—We thank Dr. B. de Winter and Miss M. Welman, National Botanic Research Institute, Pretoria, for their help and for identification of plant material, and the Deutsche Forschungsgemeinschaft for financial support.

### REFERENCES

- Merxmüller, H., Leins, P. and Roessler, H. (1977) in *The Biology and Chemistry of the Compositae* (Heywood, V. H., Harborne, J. B. and Turner, B. L., eds.), p. 577. Academic Press, New York.
- Leins, P. (1971) *Bot. J.* **691**, 91.
- Parra, O. and Marticorena, C. (1972) *Gayana* **21**, 1.
- Skvarla, J. J., Turner, B. L., Patel, V. C. and Tomb, A. S. (1977) in *The Biology and Chemistry of the Compositae* (Heywood, V. H., Harborne, J. B. and Turner, B. L., eds.), p. 164. Academic Press, New York.
- Klein, E. and Schmidt, N. (1973) *Dragoco Rep.* **20**, 3.
- Brooks, C. J. W. and Campell (1969) *J. Chem. Soc. D* **12**, 630.
- Bohlmann, F. and Suwita, A. (1979) *Phytochemistry* **18**, 677.
- Rustaiyan, A., Nazarians, L. and Bohlmann, F. (1979) *Phytochemistry* **18**, 883.
- Drozd, B., Samek, Z., Holub, M. and Herout, V. (1968) *Collect. Czech. Chem. Commun.* **33**, 1730.
- Suchy, M., Samek, Z., Herout, V. and Sorm, F. (1967) *Collect. Czech. Chem. Commun.* **32**, 2016.
- Bohlmann, F. and Zdero, C. (1980) *Phytochemistry* **19**, 149.
- Mathur, S. B., Hiremath, S. V., Kulkarni, G. H., Kelkar, G. R. and Bhattacharyya, S. C. (1965) *Tetrahedron* **21**, 3575.
- Romo de Vivar, A., Cabrera, A., Ortega, A. and Romo, J. (1967) *Tetrahedron* **23**, 3703.
- Joshi, B. S., Bawdekar, A. S., Kulkarni, G. H., Rao, A. S., Kelkar, G. H. and Bhattacharyya, S. C. (1962) *Essent. Oil Rec.* **52**, 773.
- Bohlmann, F. and LeVan, N. (1977) *Phytochemistry* **16**, 487.
- Bohlmann, F., Zdero, C., Robinson, H. and King, R. M. (1981) *Phytochemistry* **20**, 739.
- Bermejo Barrera, J., Bretton-Funes, J. L., Fajardo, M. and Gonzales, A. G. (1967) *Tetrahedron Letters* 3475.
- Bohlmann, F., Zdero, C., Robinson, H. and King, R. M. (1981) *Phytochemistry* **20**, 1631.

19. Bohlmann, F. and Zdero, C. (1979) *Phytochemistry* **18**, 95.
20. Bohlmann, F. and LeVan, N. (1978) *Phytochemistry* **17**, 570.
21. Bohlmann, F., Jakupovic, J., Robinson, H. and King, R. M. (1981) *Phytochemistry* **20**, 109.
22. Tomassini, T. C. B. and Gilbert, B. (1972) *Phytochemistry* **11**, 1177.
23. Nagumo, S., Izawa, K., Higahiyama, K. and Nagai, M. (1980) *Yakugaku Zasshi* **100**, 427.
24. Vichnewski, W., Herz, W. and Kumar, N. (1979) *J. Org. Chem.* **44**, 2575.